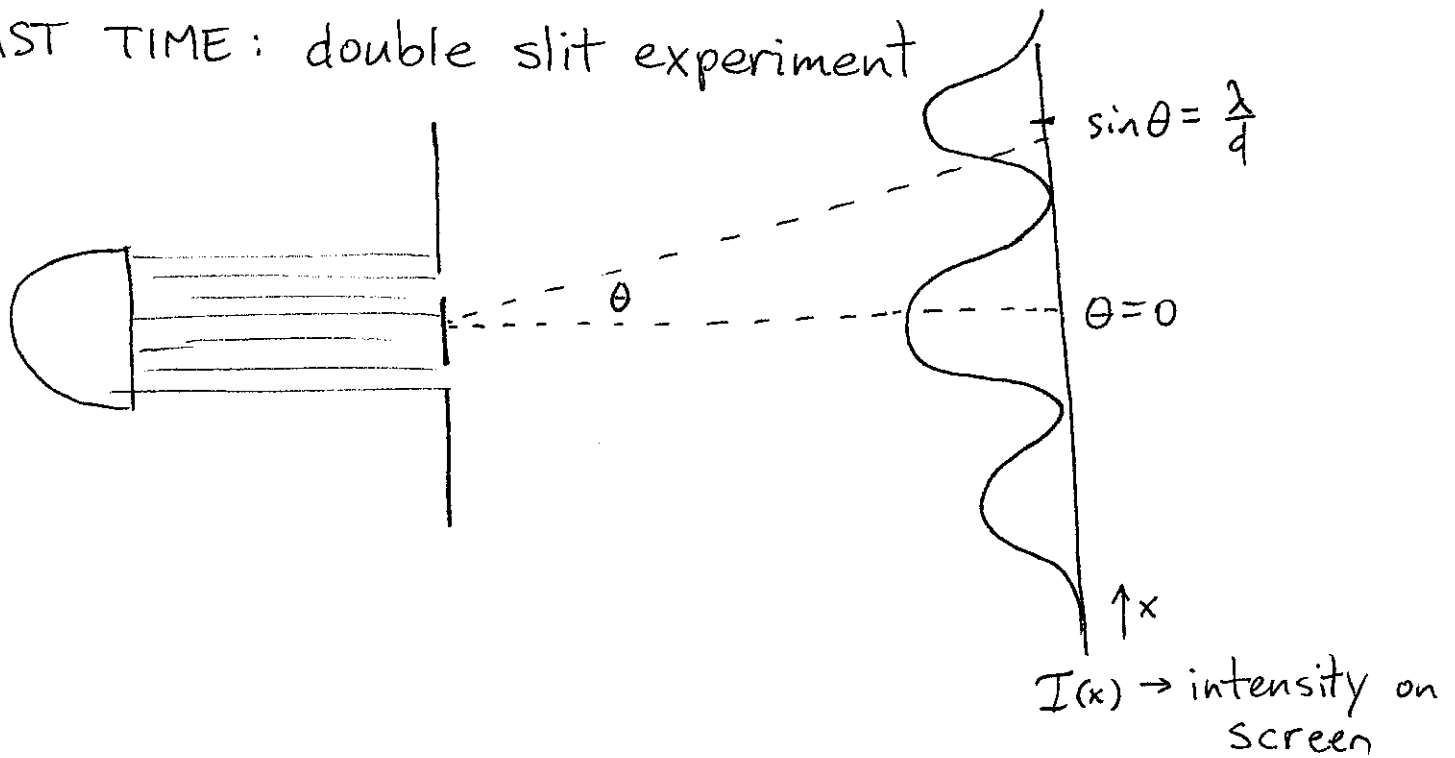


LAST TIME: double slit experiment



single photon picture:

\rightarrow each photon hits screen at specific spot
with probability density $P(x) \propto I(x)$

2 slits essential \rightarrow each photon spread out during
experiment (sees both slits)

BUT: has specific location when we
measure it.

Explanation: QUANTUM SUPERPOSITION

state of photon in flight = superposition of states
w. definite location.
("position eigenstates")

when photon hits screen (measurement) \rightarrow - becomes one of the eigenstates
- randomly "chooses" a position
w. prob. given by squared coeffs
in superposition.

Surprise: same results with electrons!

Get diffraction pattern corresponding to

$$\lambda = \frac{h}{|\vec{p}|}$$

DE BROGLIE WAVELENGTH

e.g. 100 eV $\lambda \approx 10 \text{ nm}$

→ quantum description of light & matter essentially the same.

Mathematical description of quantum superpositions

recall: all polarization states superpositions of 2 eigenstates

$$|\theta\rangle = a_{0^\circ} |0^\circ\rangle + a_{90^\circ} |90^\circ\rangle$$

↑ choice of basis states

$$|a_{0^\circ}|^2 + |a_{90^\circ}|^2 = 1$$

position: one eigenstate (basis vector) for every point in space $|\vec{x}\rangle = |(x, y, z)\rangle$

more general: $a_{\vec{x}_1} |\vec{x}_1\rangle + a_{\vec{x}_2} |\vec{x}_2\rangle + \dots$

most general state:

$$\int d^3\vec{x} \psi(\vec{x}) |\vec{x}\rangle$$

↑
integral
= "sum" over
all possible
positions

↑
coefficient
in superposition.

If we measure position:

$|\psi(x)|^2$ gives probability density for finding electron at \vec{x}

e.g.



prob. that we will find electron in volume V

$$= \int_V d^3x |\psi(\vec{x})|^2$$

(just like finding mass from mass density)

Net probability must be 1 \therefore require $\int_{\text{all space}} d^3x |\psi(x)|^2 = 1$

$\psi(\vec{x}) = \text{WAVEFUNCTION}$