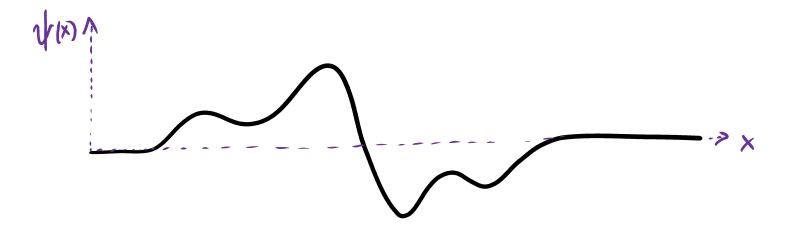
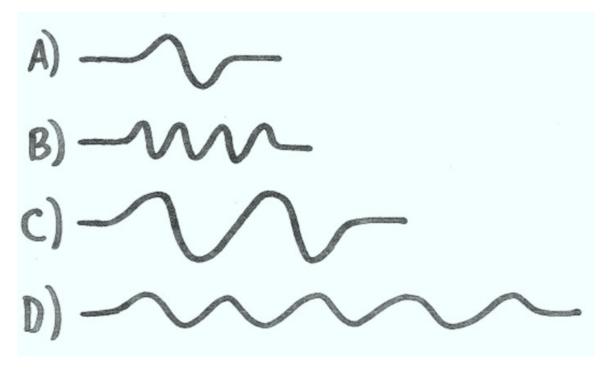
At time t=0, the wavefunction for a particle is shown below.



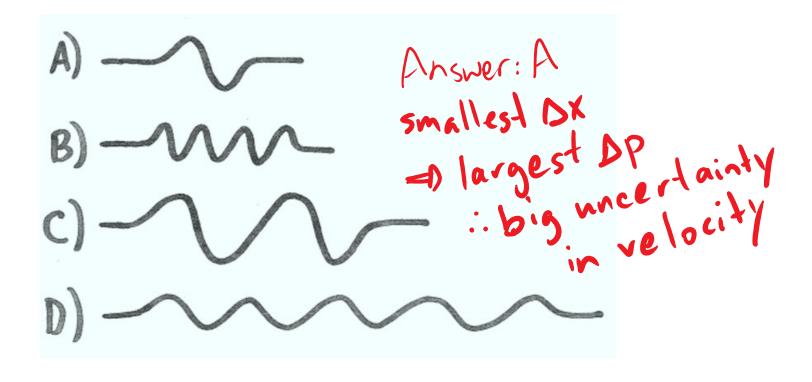
How can we determine (in principle) the wavefunction at a later time t=T?

Wavepackets for travelling particles tend to spread out as they move. Which of the wavepackets below will spread out the

fastest?



Which of the wavepackets below will spread out the fastest?



So FAR: Pretty boring physics

<u>→</u>

wave parkets travel & spread.

What if we have forces?

e.g. Coulomb force of proton in H atom.

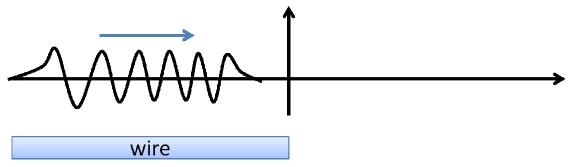
U(x)

describe by potential U(x)

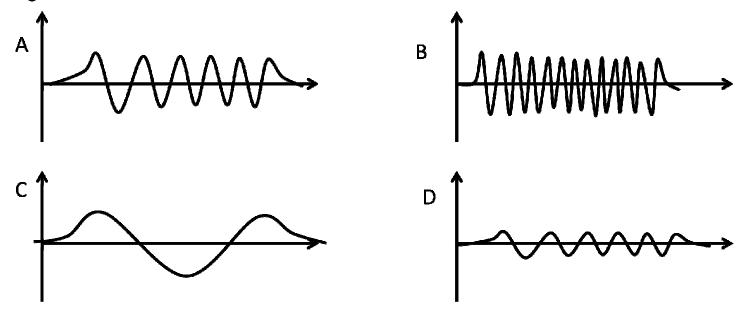
Simple example: locally constant potential

inside wire: outside wire: higher potential energy lower potential energy

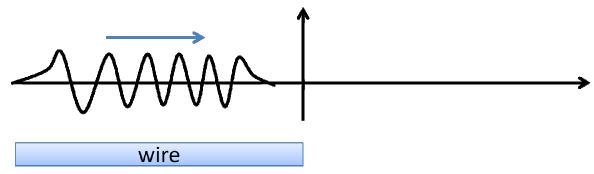
## CLICKER QUESTION:



An electron travelling along a wire with a nearly definite momentum has just enough energy to escape from the wire. If the wavefunction just before the electron exits the wire is shown above, which of the following might represent the wavefunction in the region x>0 outside the wire sometime later?

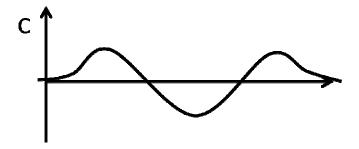


Extra: what can we say about the frequency of the wave in the x > 0 region?



An electron travelling along a wire with a nearly definite momentum has just enough energy to escape from the wire. If the wavefunction just before the electron exits the wire is shown above, which of the following might represent the wavefunction in the region x>0 outside the wire sometime later?





By energy conservation, the electron must have less kinetic energy in the region where the potential is larger. Thus, its momentum will be smaller, so its wavelength will be larger here.

Extra part: the frequency should be the same everywhere, since frequency corresponds to total energy (or since the oscillation in the x>0 region is caused by the oscillation in the x<0 region, so the frequencies should match).

Lesson:

In regions with larger U:

- Same frequency (since same total energy)
- larger wavelength (since smaller momentum)

Still have:
$$P = \frac{h}{\lambda} \quad \text{and} \quad hf = E = \frac{1}{2}mv^2 + \mathcal{U} = \frac{p^2}{2m} + \mathcal{U}$$

$$\Rightarrow f = \frac{1}{h} \left( \frac{h^2}{2m\lambda^2} + \mathcal{U} \right)$$
Sinusoidal waves move with  $v = \lambda \cdot f$ 

- Evolution for general wavefunctions by superposition

This is equivalent to a differential eqn:

$$\frac{d\psi}{dt} = (number) \cdot \frac{1}{m} \frac{d^2\psi}{dx^2} + \frac{1}{m} \frac{\psi}{dx^2} + \frac{$$

A Correct equ. even when U depends on X &

This is the full SCHRÖDINGER EQUATION

= Newton's 2nd Law for wavefunctions

replaces: 
$$\frac{dv}{dt} = \frac{F}{m}$$
 $\frac{dx}{dt} = v$ 

Sim	iulate	. Via d	computer:				
_ UK)	Find	that	valley in	potential	Can	"trap"	wavefunction
			1		4		

- trapped wavefunctions oscillate w. Specific frequencies: standing waves.

interprotation: bound particles have specific "QUANTIZED" energies

WHY: Confined waves have specific frequencies/wavelengths same as guitar string/kelp horn.

Which wavefunctions simply oscillate with frequency five. correspond to particles with energy E?

Look for solution

 $\psi(x,t) = \Psi(x) \times \text{oscillating function of time}$ 

Works if  $t^2 d^2 \overline{\psi}$ 

 $-\frac{t^2}{2m}\frac{d^2\Psi}{dx^2} + U(x)\overline{\Psi} = E\overline{\Psi}$ 

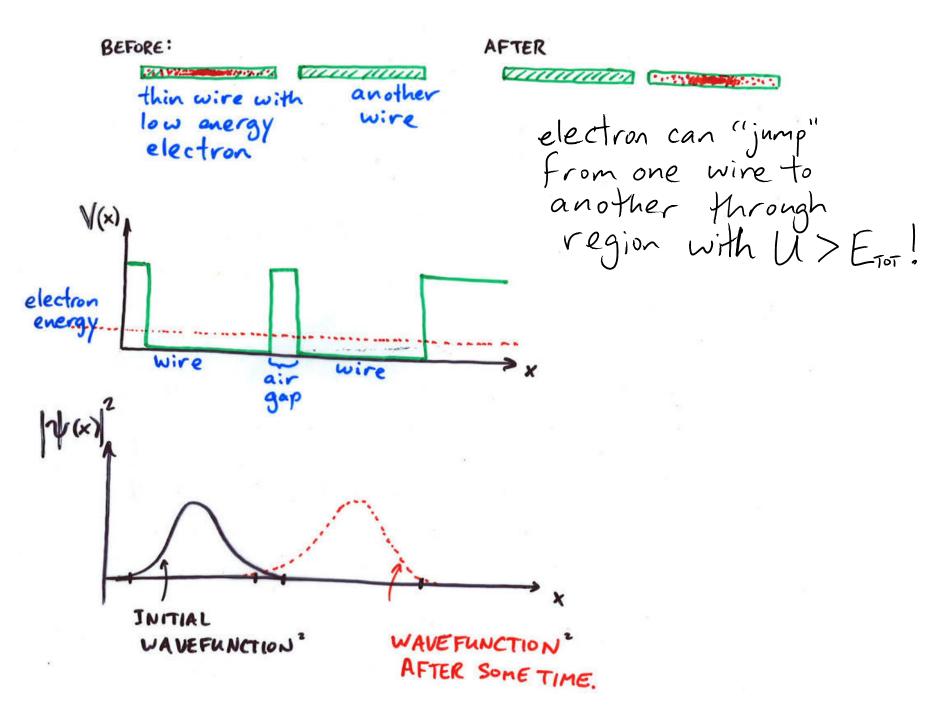
This is the "TIME-INDEPENDENT SCHRÖDINGER EQUATION"
(that you saw in chem.)



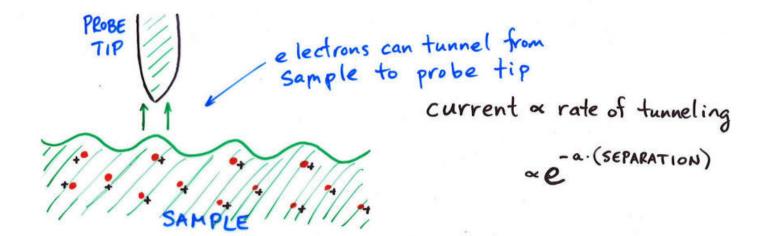
Cool fact: wavefunctions for trapped particles are never completely confined inside region where E < U.

Small probability of finding particle where it doesn't have enough energy to be.

DRAMATIC CONSEQUENCE: TUNNELING



## SCANNING - TUNNELING MICROSCOPES



Move tip horizontally across surface + measure current to deduce height profile

